Asymmetric syntheses of isostegane derivatives using trivalent iodine reagents

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Two asymmetric syntheses of the isostegane derivative (+)-3 are reported, both involving oxidation of a homochiral 2,3-dibenzylbutyrolactone with phenyliodonium bis(trifluoroacetate). Of particular interest is the fact that the oxidative step can be carried out directly on a precursor containing a hemiacetal functional group and leads to the same configuration of the biaryl unit.

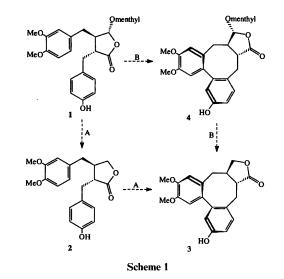
We have previously shown that phenyliodonium diacetate (PIDA) and phenyliodonium bis(trifluoroacetate) (PIFA) can be used to bring about two-electron oxidation of phenols leading to quinone-monoketals and cyclohexa-2,5-dienones.^{1.2} We have also shown that in the absence of an external nucleophile these reagents can be used to bring about intramolecular oxidative coupling leading, in the case of 2,3-dibenzylbutyrolactones, to spirodienones and dibenzocyclo-octadiene lignans.^{3.4} We have now adapted these reactions to provide an asymmetric synthesis of such compounds by starting with the homochiral 2,3-dibenzylbutyrolactone (-)-1.⁵

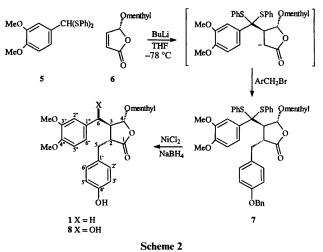
We envisaged two possible routes from 1 to compounds of the dibenzocyclooctadiene series (Scheme 1). The first (route A) would involve removing the menthyloxy group from 1 to give the homochiral 2,3-dibenzylbutyrolactone 2. We have previously shown that treatment of racemic 2 with PIFA gives the isostegane derivative 3.^{3.4} The second route (route B) would involve treating 1 directly with PIFA which would be expected to give the menthyloxy substituted dibenzocyclooctadiene 4. Removal of the menthyloxy substituent from 4 would then give the parent dibenzocyclooctadiene 3 (or a stereoisomer).

The 2,3-dibenzylbutyrolactone (-)-1 was prepared using the tandem conjugate addition methodology developed in these and other laboratories.⁴⁻⁸ Thus, conjugate addition of the lithio derivative of 3,4-dimethoxybenzaldehyde diphenyl dithio-acetal 5 to (-)-(4R)-4-menthyloxybut-2-enolide 6,⁹⁻¹¹ followed by *in situ* trapping with 4-benzyloxybenzyl bromide, gave a single adduct (-)-7 in 61% yield (Scheme 2). The relative stereochemistry at C-2 and C-3 was assigned by analogy with other tandem conjugate addition products,³⁻⁶ and from the lack of coupling between H-2, H-3 and H-4 in the ¹H NMR spectrum. Treatment of (-)-7 with nickel boride^{4.6.12.13} gave (-)-1 in 60% yield, along with 9% of the benzylic alcohol (-)-8, which was obtained as a mixture of two epimers. The ¹H and ¹³C NMR spectra of (-)-7, (-)-1 and (-)-8 are listed in Tables 1 and 2, respectively.

Treatment of (-)-1 with PIFA in trifluoroethanol (TFE) for 24 hours gave a mixture of products which were purified by chromatography to give the isostegane derivatives (+)-4 and (+)-9 in 44% combined yield, along with minor amounts (5 and 2% respectively) of the stegane isomer (-)-10 and the trifluoroethoxy derivative 11 (Scheme 3). The intermediate spirodienone 12 was not isolated but HPLC analysis indicated that after 0.5 h an intermediate, for which this is the most likely structure, was the major product present. Such compounds have been isolated and fully characterised in related reactions.^{3,4}

The structures of (+)-4, (+)-9 and (-)-10 were based on an analysis of their ¹H and ¹³C NMR spectra (Tables 3 and 4) and, in particular, upon comparison with the spectra of similar racemic compounds lacking only the menthyloxy substituent.^{3.4.6} Thus, the isostegane derivatives (+)-4 and (+)-9





showed characteristic signals in their ¹³C NMR spectra at δ 46.6 and 50.6, and 47.6 and 50.8, respectively, due to C-6 and C-7, while (–)-10 had corresponding signals at δ 41.9 and 47.5, characteristic of the stegane series.^{3,4,14,15} That 4 and 9 only differed in their configuration at C-3 was evident from their ¹H NMR spectra, which differed only in that they contained doublets at δ 5.51 (*J* 4.70 Hz) and 5.23 (*J* 8.15 Hz) respectively.

It is assumed that epimerisation at C-3 is brought about by trifluoroacetic acid formed during the reaction. It is of interest to note that both (+)-4 and (+)-9 arise from an aryl migration from the spirodienone 12, in line with our previous experience using related compounds.^{3.4}

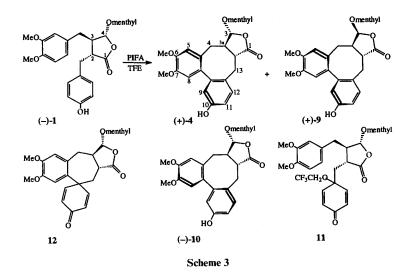


Table 1 ¹H NMR spectra of dibenzylbutyrolactones 7, 1, 8, 11, 2 and diol 15^{*a.b*}

	Peak position, multiplicity (J values/Hz) ^c							
Proton	7	1	8 ^d	11	2	15		
H-1				_		3.46-3.75m		
H-2	3.34m	2.56m	2.41m	2.59m 🔪	2.47–2.62m	1.89br s		
H-3	3.07br s	2.31m	2.33m	2.32m ∫	2.47-2.02m	1.0901 3		
H-4	5.73br s	5.39d (1.4)	5.87s	4.50s	4.11dd (6.5, 9.1) 3.88m	3.46-3.75m		
H-5	2.94-3.03m	3.01dd (4.8, 13.5)	2.94dd (4.5, 13.5)	2.02m	2.91m			
		2.75dd (10.2, 13.5)	2.68dd (11.0, 13.5)	2.19m		2.63-2.76m		
H-6		2.71m 2.40d (9.5)	4.27d (9.2)	2.67m 2.40d (10.4)	2.47–2.62m			
H-2″	7.11–7.44m	6.32d (1.8)	6.43d (1.5)		6.47d (1.8)	6.63d (1.8)		
H-5″	6.80d (8.8)	6.71d (8.1)		6.72–6.84m	6.76d (8.2)	6.76d (8.0)		
H-6″	7.06dd (1.9, 8.8)	6.50dd (1.8, 8.1)			6.57dd (1.8, 8.2)	6.68dd (1.8, 8.0)		
H-2'/6'	6.96d (8.5)	6.79d (8.5)	6.27–6.29m	6.34dd (1.6, 10.6)	6.97d (8.5)	6.95d (8.4)		
H-3'/5'	6.67d (8.5)	6.66d (8.5)		6.28dd (1.6, 10.6)	6.77d (8.5)	6.70d (8.4)		
OMe	3.67s	3.73s	3.67s	3.88s	3.79s	3.79s		
	3.84s	3.86s	3.85s	3.88s	3.83s	3.83s		
OCH,Ph	5.02s							
OCH, Ph	7.11–7.44m	<u> </u>						
OH		6.25s	6.20s		6.75s			
OCH_2CF_3				3.62m	—			

^{*a*} Samples dissolved in CDCl₃. ^{*b*} Omenthyl and SPh groups not included. ^{*c*} Multiplicities: s = singlet, d = doublet, dd = doublet of doublets, m = multiplet, br = broad. ^{*d*} Only major isomer listed.

Table 2 13 C NMR spectra of dibenzylbutyrolactones 7, 1, 8, 2 and diol $15^{a,b,c}$

Carbon	7	1	8 ^d	2	15
C-1	177.4	178.4	178.8/178.7	179.6	60.2
C-2	47.9	46.9	47.7	46.5	44.0
C-3	45.6	46.4	42.2/44.8	41.1	44.0
C-4	100.1	103.7	101.9/101.9	71.6	60.4
C-5	37.2	35.9	35.9/ 35.1	34.0	35.2
C-6	71.2	37.4	72.7/ 74.3	38.0	35.7
C-1″	130.4	129.6	133.6/133.4	130.5	133.2
C-2″	110.1	111.0	108.5	111.3	111.2
C-3″	148.6	148.8	149.0/149.2	148.9	148.8
C-4″	148.2	147.6	148.5/148.8	147.6	147.2
C-5″	112.7	111.7	110.9/110.4	111.7	112.1
C-6″	120.7	121.2	117.6/119.2	120.7	121.1
C-1′	130.7	129.9	129.4	129.0	131.9
C-2'/6'	130.7	130.0	130.2/130.0	130.4	130.0
C-3'/5'	114.7	115.4	115.2/115.4	115.5	115.3
C-4'	157.6	154.5	154.4/154.5	155.1	154.4
OMe	55.7	55.5	55.6/ 55.5	55.8	55.8
OMe	55.7	55.9	56.0	55.9	55.9

^a Samples dissolved in CDCl₃. ^b OCH₂Ph, Omenthyl and SPh groups not included. ^c Assignments supported by DEPT spectra. ^d Both isomers included.

Treatment of a mixture of (+)-4 and (+)-9 with NaBH₄ and KOH^{4,6} gave as a major product (+)-3 (37%) in which the menthyloxy group had been removed, along with smaller amounts of the diol (+)-13 (28%) and the hydroxy acid 14 (22%) (Scheme 4). When compound 14 was left in a desiccator over P₂O₅ it was slowly and quantitatively transformed into the lactone (+)-3. Thus the overall yield of (+)-3 from 4 and 9 was 59%. The structure of (+)-3 was firmly established by comparison of its ¹H and ¹³C NMR spectra (Tables 3 and 4) with those of the corresponding racemic compound.^{3.4}

Finally, treatment of the 2,3-dibenzylbutyrolactone (-)-1 with NaBH₄ and KOH gave as the major product (-)-2 (43%) along with the diol (+)-15 (11%). The ¹H and ¹³C NMR spectra of (-)-2 and (+)-15 are listed in Tables 1 and 2. Treatment of (-)-2 with PIFA in TFE gave once again the isostegane (+)-3, identified by HPLC comparison with the sample prepared from (+)-4 and (+)-9 (Scheme 5).

The methodology described in this paper therefore provides an alternative to the known methods for the synthesis of dibenzocyclooctadienes by oxidative aromatic coupling,^{7,8,16,17} and illustrates the application of this methodology to two alternative approaches to the asymmetric synthesis of such compounds.

Table 3	¹ H NMR spectra	of dibenzocyc	clooctadiene	derivatives a.b
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	Peak position, multiplicity (J values/Hz) ^c							
Proton	4	9	10	3	13 ^d	14 ^{<i>d</i>}		
H-4	2.69d (13.3) 2.60dd (9.7, 13.3)	2.77d (12.9) 2.35dd (9.7, 12.9)	2.93m 2.83m	2.64d (13.2) 2.38dd (9.8, 13.2)	2.82d (13.3) 2.04–2.11m	2.72d (13.5) 2.15–2.42m		
H-3a H-13a }	2.16–2.26m }	2.15-2.26m	2.18–2.27m 2.33–2.41m	2.20m 2.10dd (9.4, 13.2)	· 1.53br s }	1.96-2.07m		
H-13	2.49dd (9.9, 13.6) 3.21d (13.6)	2.30m 3.23d (12.5)	2.76m 3.02dd (2.6, 10.9)	2.26dd (9.4, 13.6) 3.17d (13.6)	2.04–2.11m 2.83d (13.2)	2.15–2.42m 2.78d (13.4)		
H-3	5.51d (4.7)	5.23d (8.1)	5.27d (5.1)	4.39dd (6.7, 8.4) 3.77dd (11.2, 8.4)	3.74–3.78m 3.33–3.40m	2.63dd (2.6, 10.2) 3.25–3.37m		
H-8	6.67s	6.67s	6.70s	6.66s	6.78s	6.76s		
H-5	6.68s	6.68s	6.58s	6.66s	6.89s	6.92s		
H-12	7.18d (8.3)	7.14d (8.4)	6.73d (6.4)	7.13d (8.3)	7.13d (7.9)	7.18d (8.2)		
H-11 H-9	6.86dd (2.7, 8.3) 6.70d (2.7)	6.83dd (2.8, 8.4) 6.67d (2.8)	6.98dd (2.4, 6.4) 6.74d (2.4)	6.81dd(2.7, 8.3) 6.68d(2.7)	6.73m	6.80dd (2.5, 8.2) 6.73d (2.5)		
OMe	3.84s 3.95s	3.84s 3.94s	3.84s 3.94s	3.90s 3.81s	3.82s 3.82s	3.82s 3.85s		
ОН	5.92s	5.99s	4.99s	6.06br s	3.04s	5.57s		
H-1				—	3.74–3.78m 3.33–3.40m			

^a Samples dissolved in CDCl₃, except where indicated. ^b Omenthyl signals not listed. ^c For multiplicities, see Table 1 footnote (c). ^d Sample dissolved in $[^{2}H_{6}]$ acetone.

Table 4 ¹³C NMR spectra of dibenzocyclooctadiene derivatives^{*a.b.c.*}

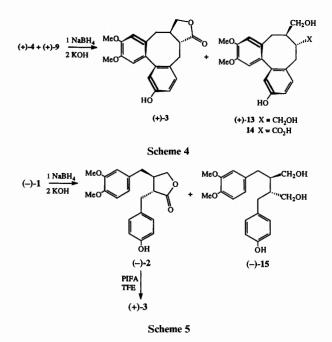
Carbon	4	9	10	3	13 ^d	14 ^d
C-1	177.2	174.3	178.7	177.2	66.6	177.4
C-3	104.8	103.1	104.3	70.2	66.6	66.2
C-3a	46.6	46.5	41.9	46.7	45.2	45.8
C-4	34.1	34.2	32.3	34.1	35.6	35.1
C-4a	132.5	132.4	132.6	131.8	133.8	133.2
C-5	114.1	113.9	114.0	114.0	113.7	113.5
C-6	148.7	148.7	147.8	148.7	149.8	149.9
C-7	147.1	147.1	147.1	147.1	148.2	148.2
C-8	112.2	112.4	113.8	112.0	112.6	112.7
C-8a	131.9	131.7	129.2	132.4	134.5	133.9
C-8b	131.0	130.5	129.1	130.5	133.5	131.1
C-9	115.3	115.3	114.4	115.4	115.2	115.4
C-10	154.3	154.3	154.5	154.3	155.9	156.3
C-11	117.5	117.6	117.0	117.7	115.5	115.7
C-12	130.5	130.3	130.4	130.4	131.0	131.0
C-12a	141.8	141.8	144.1	141.8	142.4	142.5
C-13	33.4	32.8	34.2	31.4	35.2	33.7
C-13a	50.6	50.8	42.5	50.5	45.8	49.7
OMe	56.0	56.0	55.9	56.0	56.1	56.0
	56.1	55.1	55.9	56.0	56.2	56.0

^a Samples dissolved in CDCl₃ unless otherwise indicated. ^b Omenthyl signals not listed. ^c All assignments are consistent with DEPT spectra. ^d Samples dissolved in [²H₆]acetone.

Experimental

IR spectra were recorded on a Pye Unicam SP1050 spectrometer. UV spectra were recorded on a Philips PU8720 scanning spectrometer. ¹H NMR spectra were recorded on a Bruker 250WM spectrometer at 250 MHz. The high field spectra were recorded using a Bruker AC 400 spectrometer at 400 MHz. ¹³C NMR spectra were recorded on a Bruker 250WM spectrometer at 62.9 MHz. All NMR spectra used tetramethylsilane as the internal standard, and were run in deuteriated chloroform, unless otherwise stated. The mass spectra were recorded on a VG-12-250 low resolution quadruple mass spectrometer, whilst accurate mass measurements were obtained from a ZAB-E, high resolution, double focussing mass spectrometer. Melting points were recorded on an Electrothermal digital melting point apparatus, and are uncorrected.

Analytical HPLC work was carried out on a Milton Roy instrument, consisting of a 3100 SpectroMonitor, 3000 ConstaMetric pump, and CI-4100 integrator, and used an Apex



II ODS 5 μ m column. Thin layer chromatography (TLC) was carried out on Merck 5735 Kieselgel 60 F₂₅₄ fluorescent plates. Flash chromatography was performed with silica gel (Merck 9385, Kieselgel 60, 230–400 mesh) or neutral alumina. Small scale purifications were conducted on a Chromatotron 7924 using 1 mm, 2 mm or 4 mm plates prepared from silica gel (Merck 7749, Kieselgel 50 F₂₅₄ gipshaltig).

Reactions carried out under an inert atmosphere refer to the use of argon or 'white spot' nitrogen used directly from the cylinder. Tetrahydrofuran (THF) was dried by being stirred overnight over calcium hydride, passed down a dry alumina column, and then distilled from sodium wire and benzophenone. Dichloromethane was dried by passing down a dry alumina column and then distilled from calcium hydride. Methanol and ethanol were dried by treatment with magnesium activated with iodine followed by heating under reflux and distillation. Solutions of butyllithium in hexane were obtained from Aldrich and were regularly estimated.^{18,19} Low temperature baths were prepared by making a slurry of solid carbon dioxide with acetone (-78 °C) or using a mixture of solid carbon dioxide with acetone monitored using a digital temperature probe (-25

to -30 °C). All other reagents were purified by distillation under nitrogen prior to use, the pressure being reduced if the boiling point of the compound was >110 °C at atmospheric pressure.

Preparation of $(-)-(2R,3R,4R)-2-(4'-benzyloxybenzyl)-3-{3",4"-dimethoxyphenyl[bis(phenylsulfanyl)]methyl}-4-(menthyloxy)butyrolactone 7$

The diphenyldithioacetal 5²⁰ (3.00 g, 8.15 mmol) was dissolved in dry THF (50 ml), under an argon atmosphere, and cooled to -78 °C. To this stirred solution was added, via syringe, BuLi (2.94 mol 1⁻¹; 3 ml, 8.8 mmol, 1.08 mol equiv.) and stirring was continued at -78 °C for 2.5 h. After this time, pre-cooled (-)-4-menthyloxybut-2-enolide 6⁶ (2.6 g, 10.92 mmol, 1.34 mol equiv.), dissolved in dry THF (16 ml), was added via a doubleended needle. Stirring was continued at -78 °C for 2.5 h before pre-cooled 1,3-dimethylimidazolidin-2-one (DMI) (1 ml, 1.04 g, 9.11 mmol, 1.12 mol equiv.) was added via syringe, immediately followed by pre-cooled 4-benzyloxybenzyl bromide²¹ (2.5 g, 9.03 mmol, 1.11 mol equiv.) dissolved in dry THF (11 ml), which was added via a double-ended needle. The mixture was stirred, and allowed to warm to room temperature overnight, before adding water (60 ml) and extracting with EtOAc (3×100 ml). The combined organic extracts were washed with aqueous NaCl $(3 \times 50 \text{ ml})$, dried (MgSO₄), filtered and evaporated, vielding a thick reddish oil. Purification by flash chromatography on silica and elution with light petroleum (bp 60-80 °C)-CH₂Cl₂ (85:15) afforded (-)-7 as a white foam (3.97 g, 60.7%), mp 50–55 °C, $[\alpha]_{D}^{22}$ – 103.38 (*c* 0.207, CHCl₃) (Found: C, 73.5; H, 7.3. C₄₉H₅₄O₆S₂ requires C, 73.32; H, 6.73%); $v_{max}(film)/cm^{-1}$ 1779 (γ -lactone); see Tables 1 and 2 for ¹H and ¹³C NMR data; m/z (EI) 584 (M - 2SPh, 83%), 151 $(Ar^{1}CH_{2}^{+}, 10\%)$, 91 $(PhCH_{2}^{+}, 100\%)$; m/z (CI) 693 (M - m)SPh, 3%), 585 (M + H - 2SPh, 10%), 259 (Ar¹CHSPh⁺, 100%) [Found (HRMS): $(M - 2SPh)^+$, 584.3138. $C_{37}H_{44}O_6$ requires $(M - 2SPh)^+$, 584.3138].

Preparation of (-)-(2R,3R,4R)-2-(4'-hydroxybenzyl)-3-(3'',4''-dimethoxybenzyl)-4-(menthyloxy)butyrolactone 1 and <math>(-)-(2R,3S,4R)-2-(4'-hydroxybenzyl)-3-[3'',4''-dimethoxy-phenyl(hydroxy)methyl]-4-(menthyloxy)butyrolactone 8

To (-)-7 (1.14 g, 1.42 mmol), dissolved in MeOH (130 ml) and THF (44 ml) was added NiCl₂•6H₂O (96.77 g, 28.48 mmol, 20.06 mol equiv.). The stirred green solution was cooled to 0 °C and NaBH₄ (3.23 g, 85.36 mmol, 60.11 mol equiv.) was added carefully, in order to minimize the effervescence produced. The mixture was removed from the ice-bath and stirred for 1 h at room temperature. After this time, water (20 ml) was added and the reaction mixture passed through a short Celite column, in order to remove the nickel salts. Water (100 ml) was added and the resulting solution extracted with $CHCl_3$ (3 × 150 ml). The combined organic extracts were dried (MgSO₄), filtered and evaporated to give a colourless gum. Purification by flash chromatography on silica and elution with light petroleum (bp 60-80 °C)-EtOAc (50:50) afforded (-)-1 as a white crystalline foam (0.43 g, 60%), mp 50–55 °C. $[\alpha]_D^{22}$ –145.8 (c 0.402, CHCl₃) (Found: C, 72.5; H, 8.2. C₃₀H₄₀O₆ requires C, 72.58; H, 8.06%); $v_{max}(film)/cm^{-1}$ 3420 (OH), 1768 (γ -lactone); $\lambda_{max}(Me-1)$ OH)/nm 209.5 (ɛ/l mol⁻¹ ml⁻¹ 16 850), 227.2 (15 810), 280.3 (5014); see Tables 1 and 2 for ¹H and ¹³C NMR data; m/z (EI) 496 (M⁺, 4%), 357 (M - menthyl, 7), 251 (31), 151 (Ar¹CH₂⁺, 92), 107 ($\operatorname{Ar}^2\operatorname{CH}_2^+$, 100); m/z (CI) 514 (M + NH₄⁺, 14%) 497 (M + H⁺, 13), 358 (M + H - menthyl, 6), 341 (M -Omenthyl, 100) (Found: M⁺, 496.2825. C₃₀H₄₀O₆ requires M, 496.2825).

Further elution with light petroleum (60–80 °C)–EtOAc (65:35) gave (–)-**8** as a white gum (0.66 g, 9%), $[\alpha]_D^{22}$ –115.57 (*c* 0.244, CHCl₃); ν_{max} (film)/cm⁻¹ 3446 (OH), 1747 (γ-lactone); see Tables 1 and 2 for ¹H and ¹³C NMR data; *m/z* (EI) 512 (M⁺, 13%), 356 (M – menthol, 15), 250 (4), 167 (Ar¹CHOH⁺, 45),

107 ($Ar^{2}CH_{2}^{+}$, 100); m/z (CI) 530 (M + NH₄⁺, 4%), 374 (M + H - menthyl, 25), 357 (M - Omenthyl, 44), 167 ($Ar^{1}CHOH^{+}$, 100) (Found: M⁺, 512.2770. C₃₀H₄₀O₇ requires M, 512.2774).

Preparation of (+)-(3R,3aR,8a/8bS,13aR)-10-hydroxy-3-(menthyloxy)-6,7-dimethoxy-3a,4,13,13a-tetrahydro-1*H*,3*H*dibenzo[4,5:6,7]cycloocta[1,2-*c*]furan-1-one 4 and (+)-(3S,3aR,8a/8bS,13aR)-10-hydroxy-3-(menthyloxy)-6,7dimethoxy-3a,4,13,13a-tetrahydro-1*H*,3*H*-dibenzo[4,5:6,7]cycloocta[1,2-*c*]furan-1-one 9

To (-)-1 (0.11 g, 0.22 mmol) dissolved in dry trifluoroethanol (TFE) (2 ml), under an argon atmosphere, was added, phenyliodonium bis(trifluoroacetate) (PIFA) (0.11 g, 0.256 mmol, 1.16 mol equiv.) dissolved in dry TFE (2 ml) via syringe. Stirring was continued at room temperature for 24 h, after which time, the reaction mixture was neutralized by addition of powdered NaHCO₃, the mixture concentrated in vacuo, and the residue dissolved in EtOAc and filtered. The filtrate was evaporated and the residue purified by flash chromatography on silica using light petroleum (bp 60-80 °C)-EtOAc (80:20) which yielded (+)-4 together with (+)-9 (0.048 g, 44%) as a white gum. Some (+)-4 was separated by a second column; $[\alpha]_{D}^{22}$ +186.9 (c 0.312, CHCl₃); $v_{max}(film)/cm^{-1}$ 3436 (OH), 1774 (γ-lactone); λ_{max} (MeOH)/nm 215.6 (ε/l mol⁻¹ ml⁻¹ 24 790), 280.4 (13 395); see Tables 3 and 4 for ¹H and ¹³C NMR data; m/z (EI) 494 (M⁺, 29%), 354 (M – H – menthyl, 3), 339 (M - Omenthyl, 3), 312 (17), 282 (6); m/z (CI) 512 (M + C) NH_4^+ , 1%), 495 (M + H⁺, 14), 374 (M + NH_4 + H menthyl, 100), 311 (3), 283 (4), 156 (7) [Found (HRMS): M⁺, 494.2668. C₃₀H₃₈O₆ requires *M*, 494.2668]. One fraction contained mainly isomer (+)-9; $[\alpha]_D^{22} + 84.3$ (c 0.325, CHCl₃); $\nu_{max}(film)/cm^{-1}$ 3440 (OH), 1768 (γ -lactone), see Tables 3 and 4 for ¹H and ¹³C NMR data; m/z (EI) 494 (M⁺, 57%), 354 (M – H – menthyl, 5), 339 (M – Omenthyl, 4), 312 (37), 282 (11); m/z (CI) 512 (M + NH₄⁺, 1%), 495 (M + H^+ , 14), 374 (M + NH₄ + H - menthyl, 100), 311 (9), 283 (5), 156 (7) [Found (HRMS): M⁺, 494.2668. C₃₀H₃₈O₆ requires M, 494.2668].

Continued elution with light petroleum (60-80 °C)-EtOAc (80:20)gave (-)-(3R,3aR,8a/8bR,13aR)-10-hydroxy-3-(menthyloxy)-6,7-dimethoxy-3a,4,13,13a-tetrahydro-1H,3Hdibenzo[4,5:6,7]cycloocta[1,2-c]furan-1-one 10 as a white gum $(5.5 \text{ mg}, 5\%), [\alpha]_{D}^{22} - 136.1 (c 1.00, \text{CHCl}_{3}); v_{\text{max}}(\text{film})/\text{cm}^{-1}$ 3440 (OH), 1775 (γ -lactone); see Tables 3 and 4 for ¹H and ¹³C NMR data; m/z (EI) 494 (M⁺, 100%), 354 (M - H menthyl, 12), 338 (M - menthol, 10), 282 (18), 255 (84), 242 (77); m/z (CI) 495 (M + H⁺, 3%), 374 (M + NH₄ + H menthyl, 100), 357 (M + 2H - menthyl, 73), 339 (M -Omenthyl, 27), 283 (13) [Found (HRMS): M⁺, 494.2668. $C_{30}H_{38}O_6$ requires M, 494.2668] and 2-{[1'-(2,2,2trifluoroethoxy)-4-oxocyclohexa-2',5'-dienyl)methyl}-3-(3",4"dimethoxybenzyl)-4-(menthyloxy)butyrolactone 11 as a white gum (2.65 mg, 2%); see Table 1 for ¹H NMR data; m/z (EI) 594 $(M^+, 7\%)$, 496 (M - CF₃CHO, 11), 438 (M - menthol, 6), 357 (M - menthyl - CF_3CHO , 6), 339 (M - menthol - CF_3CH_2O , 17), 151 ($Ar^1CH_2^+$, 100); m/z (CI) 514 (M + $NH_4 - CF_3CHO, 8\%$, 497 (M + H - CF_3CHO, 3), 341 $(M + H - Omenthyl - CF_3CH_2O, 31), 313$ (4), 107 (8) [Found (HRMS): M^+ , 594.2804. $C_{32}H_{41}O_7F_3$ requires *M*, 594.28041

Preparation of (+)-10-hydroxy-6,7-dimethoxy-3a,4,13,13atetrahydro-1*H*,3*H*-dibenzo[4,5:6,7]cycloocta[1,2-*c*]furan-1one 3, (+)-(6*R*,7*R*,1a/12a*S*)-6,7-bishydroxymethyl-11-hydroxy-2,3-dimethoxy-5,6,7,8-tetrahydrodibenzo[*a*,*c*]cyclooctene 13 and (6*R*,7*R*,1a/12a*S*)-6-hydroxymethyl-11-hydroxy-2,3-dimethoxy-5,6,7,8-tetrahydrodibenzo[*a*,*c*]cyclooctene-7-carboxylic acid 14 To a mixture of (+)-4 and (+)-9 (0.15 g, 0.30 mmol) dissolved in dry EtOH (5 ml), under an argon atmosphere, was added NaBH₄

(0.115 g, 3.04 mmol, 10.13 mol equiv.). KOH in EtOH (0.043 g, 1 ml of 0.75 mol 1⁻¹, 0.77 mmol, 2.57 mol equiv.) was then added, via a syringe, and stirring continued at room temperature for 5 h. The reaction was quenched by the addition of aqueous HCl until pH = 3.0. An equal volume of water (6 ml) was then added and the mixture extracted with CH_2Cl_2 (3 × 30 ml). The organic extracts were combined and the product was allowed to lactonise. This process was monitored by HPLC and typically took 24 h to reach completion. The reaction mixture was then thoroughly washed with water $(3 \times 20 \text{ ml})$, dried (MgSO₄), and filtered. Purification by flash chromatography on silica using light petroleum (bp 60-80 °C)-EtOAc (60:40) afforded (+)-3 as a white solid (0.038 g, 37%); $[\alpha]_D^{22}$ +108.8 (c 0.25, CHCl₃); mp 190–191 °C (lit.,⁴ 190–192 °C); $\nu_{max}(film)/cm^{-1}$ 3434 (OH), 1752 (γ -lactone); λ_{max} (MeOH)/nm 216.2 nm (ϵ/l mol⁻¹ ml⁻¹ 13 520), 230.9 (13 620), 281.1 (6908); see Tables 3 and 4 for ¹H and ¹³C NMR data; m/z (EI) 340 (M⁺, 100%), 325 (3), 279 (10), 255 (38); m/z (CI) 358 (M + NH₄⁺, 100%), 341 (M + H⁺, 20%) [Found (HRMS): M⁺, 340.1311. $C_{20}H_{20}O_5$ requires M, 340.1311].

Further elution with light petroleum (bp 60–80 °C)–EtOAc (10:90) gave (+)-13 as a white gum (0.029 g, 28%); $[\alpha]_D^{22}$ +63.2 (c 0.25, EtOH); v_{max} (KBr)/cm¹ 3441 (OH); λ_{max} (MeOH)/nm 214.4 (ϵ /l mol⁻¹ ml⁻¹ 37 460), 283.2 (8504); see Tables 3 and 4 for ¹H and ¹³C NMR data; m/z (EI) 344 (M⁺, 100%), 326 (M – H₂O, 6), 314 (M – CH₂O, 2), 295 (M – CH₂OH – H₂O, 12), 268 (10); m/z (CI) 362 (M + NH₄⁺, 83), 345 (M + H⁺, 64), 327 (M + H – H₂O, 100), 283 (3), 269 (6) [Found (HRMS): M⁺, 344.1624. C₂₀H₂₄O₅ requires *M*, 344.1624] and 14 as a white foam (0.02 g, 22%); see Tables 3 and 4 for ¹H and ¹³C NMR data.

Preparation of (-)-(2R,3R)-2-(4'-hydroxybenzyl)-3-(3'',4''-dimethoxybenzyl)butyrolactone 2 and <math>(-)-(2R,3R)-2-(4'-hydroxybenzyl)-3-(3'',4''-dimethoxybenzyl)butane-1,4-diol 15

To (-)-1 (0.27 g, 0.544 mmol) dissolved in dry EtOH (10 ml), under an argon atmosphere, was added NaBH₄ (0.226 g, 5.97 mmol, 10.97 mol equiv.), via a solid addition side arm. KOH in EtOH (2 ml of 0.75 mol l^{-1} , 1.51 mmol, 2.78 mol equiv.) was then added via a syringe, and stirring continued at room temperature for 4.5 h. The reaction was quenched by the addition of aqueous HCl until pH = 3.0. An equal volume of water (12 ml) was then added and the solution extracted with CH_2Cl_2 (3 × 40 ml). The organic extracts were combined and the product was allowed to lactonise. This process was monitored by HPLC and typically took 12-24 h to reach completion. The reaction mixture was then thoroughly washed with water $(3 \times 30 \text{ ml})$, dried (MgSO₄), and filtered. Purification by Chromatotron on silica using light petroleum (bp 60-80 °C)-EtOAc (50:50) afforded (-)-2 as a white solid $(0.08 \text{ g}, 43\%); \ [\alpha]_{D}^{22} - 29.5 \ (c \ 0.27, \text{ CHCl}_3); \text{ mp } 148-149 \ ^\circ\text{C}$ (lit., ⁴ 148.4–148.8 °C; $v_{max}(film)/cm^{-1}$ 3422 (OH), 1767 (γ lactone); λ_{max} (MeOH)/nm 207.4 ($\epsilon/l \text{ mol}^{-1} \text{ ml}^{-1}$ 17 970), 227.2 (13 630), 279.7 (4 416); see Tables 1 and 2 for ¹H and ¹³C NMR data; m/z (EI) 342 (M⁺, 36%), 178 (10), 151 (Ar¹CH₂⁺, 100), 107 (Ar²CH₂⁺, 73); m/z (CI) 360 (M + NH₄⁺, 91%), 343 $(M + H^+, 100)$ [Found (HRMS): M⁺, 342.1467. C₂₀H₂₂O₅ requires M, 342.1467). Further elution with light petroleum (bp 60-80 °C)-EtOAc (45:55) gave (-)-15 as a colourless gum $(0.021 \text{ g}, 11\%); [\alpha]_{D}^{22} - 32.14 (c \ 0.14, \text{CHCl}_3); v_{max}(\text{film})/\text{cm}^{-1}$ 3281 (OH); see Tables 1 and 2 for ¹H and ¹³C NMR data; m/z

(EI) 346 (M⁺, 24%), 328 (M - H₂O, 10), 300 (M - CO₂, 1), 151 (Ar¹CH₂⁺, 100), 137 (Ar¹⁺, 8), 107 (Ar²CH₂⁺, 57); *m/z* (CI) 464 (M + NH₄⁺, 36%), 347 (M + H⁺, 45), 346 (100), 329 (M + H - H₂O, 47), 311 (M + H - 2H₂O, 31), 297 (M - CH₂OH - H₂O, 3) (Found (HRMS): M⁺, 346.1780. $C_{20}H_{26}O_5$ requires *M*, 346.1780).

Preparation of (+)-10-hydroxy-6,7-dimethoxy-3a,4,13,13atetrahydro-1*H*,3*H*-dibenzo[4,5:6,7]cycloocta[1,2-*c*]furan-1one 3

The phenolic dibenzylbutyrolactone (-)-2 (0.064 g, 0.187 mmol) was dissolved in dry TFE (2 ml) under nitrogen and the solution was stirred. PIFA (0.096 g, 0.22 mmol) dissolved in dry TFE (1 ml) was then added *via* syringe and stirring was continued at room temperature for 24 h. After this time the reaction mixture was neutralised by addition of powdered NaHCO₃, the mixture concentrated *in vacuo* and the residue dissolved in EtOAc and filtered. The filtrate was examined by HPLC and shown to contain a single product which by coinjection was identical with (+)-3 prepared earlier by route B.

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